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Magnetic and Mössbauer Spectroscopic Characterization of EuNi₂Sb₂

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ThCr₂Si₂-type EuNi₂Sb₂ (I4/mmm, a=438.1(1), c=1068.1(4) pm) was synthesized by arc-melting of the elements. Magnetic susceptibility measurements show Curie-Weiss behavior with an experimental magnetic moment of 8.03(1) μ_B/Eu atom, indicating stable divalent europium. EuNi₂Sb₂ orders antiferromagnetically at $T_N=5.8(1)$ K. A 151 Eu Mössbauer spectrum at 4.2 K shows two spectral components with hyperfine fields of 19.1 and 12.9 T, indicative of different magnetically ordered domains.

Key words: Europium, Antimonide, Mössbauer Spectroscopy

Introduction

The antimonides EuT_2Sb_2 (T = Mg, Mn, Ni, Cu, Zn, Pd, Cd, Pt) [1 and refs. cited therein] crystallize with simple structures that derive from the ThCr₂Si₂, CaBe₂Ge₂ or CaAl₂Si₂ type. The europium atoms show magnetic ordering at low temperatures, and a maximum Néel temperature of 13.3 K was observed for $EuZn_2Sb_2$ [2]. Only the transition metal substructure of $EuMn_2Sb_2$ [3] orders magnetically at around 600 K.

EuNi₂Sb₂ has first been reported by Marchand and Jeitschko [4], and the structure was refined on the basis of single-crystal diffractometer data [5]. However, so far no physical properties have been determined. In the course of our systematic studies on ThCr₂Si₂-and CaBe₂Ge₂-related compounds [1, 3, 6, and refs. therein] we have determined the magnetic and Mössbauer spectroscopic behavior of this antimonide.

Experimental Section

Synthesis

Starting materials for the preparation of the $EuNi_2Sb_2$ sample were ingots of sublimed europium (Johnson

Matthey), nickel wire (Johnson Matthey) and antimony shots (ABCR), all with stated purities better than 99.9 %. The airand moisture-sensitive europium pieces were kept under argon prior to the reaction. The argon was purified before with titanium sponge (870 K), silica gel and molecular sieves. Europium, nickel, and antimony were weighed in the ideal 1:2:2 atomic ratio and reacted by arc-melting under argon [7]. The product pellet was remelted three times in order to ensure homogeneity. The total weight loss after the melting procedures was less than 0.5 %. The EuNi₂Sb₂ sample is silvery with metallic luster and stable in air over months.

Powder X-ray diffraction

The polycrystalline EuNi₂Sb₂ sample was characterized by powder X-ray diffraction: Guinier camera, $\text{Cu}K_{\alpha 1}$ radiation, α -quartz (a=491.30, c=540.46 pm) as internal standard, and an imaging plate technique (Fujifilm, BAS-READER 1800). The tetragonal lattice parameters (a=438.1(1), c=1068.1(4) pm) were deduced from a least-squares refinement. The experimental pattern was compared to a calculated one [8] to ensure correct indexing. Our data compare well with the ones reported by Marchand and Jeitschko (a=438.34(6), c=1066.4(1) pm) [4].

Susceptibility measurements

Magnetic susceptibility measurements were carried out on a Quantum Design Physical Property Measurement System (PPMS) using the VSM option. For the measurements, 7.574 mg of the EuNi₂Sb₂ sample were packed in kapton foil and attached to the sample holder rod for measuring the magnetic properties in the temperature range 2.5–300 K with magnetic flux densities up to 80 kOe.

Mössbauer spectroscopy

The 21.53 keV transition of ¹⁵¹Eu with an activity of 130 MBq (2% of the total activity of a ¹⁵¹Sm: EuF₃ source) and a Ba^{121m}SnO₃ source were used for the Mössbauer spectroscopic characterization. The measurements were conducted in transmission geometry with a commercial helium-bath cryostat. The temperature of the absorber was varied between 4.2 K and r. t., while the source was kept at r. t. The temperature was controlled by a resistance thermometer (±0.5 K accuracy). The sample thickness (PVC container) corresponds to about 10 mg of the Mössbaueractive element/cm².

Results and Discussion

Crystal chemistry

EuNi₂Sb₂ adopts the tetragonal ThCr₂Si₂-type structure, space group *I*4/*mmm*. The nickel atoms

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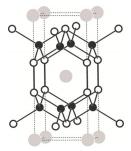


Fig. 1. View of the EuNi₂Sb₂ structure approximately along the *y* axis. Europium, nickel and antimony atoms are drawn as medium grey, black filled and open circles, respectively. The three-dimensional [Ni₂Sb₂] network is emphasized.

have tetrahedral antimony coordination (Fig. 1). A view on the crystal structure approximately along the y axis is presented in Fig. 1. The NiSb₄ tetrahedra are condensed *via* common edges to layers and *via* Sb–Sb bonds (299 pm [5]) to a three-dimensional network in which the divalent europium atoms (*vide in-fra*) fill the large cavities. The crystal chemistry of ThCr₂Si₂-related structures has repeatedly been discussed. For further details and literature overviews we refer to [6, 9, and refs. therein].

As was evident from single-crystal data [5], EuNi₂Sb₂ adopts a small homogeneity range EuNi_{2-x}Sb₂. A structure refinement revealed a composition EuNi_{1.53}Sb₂ and lattice parameters a = 434.0(1), c = 1059.7(5) pm, smaller than those originally reported for EuNi₂Sb₂ [4]. Since our lattice parameters (*vide supra*) are close to the data reported in [4], we conclude that the composition of our sample is close to the ideal one.

Magnetic properties

The $\chi(T)$ and $\chi^{-1}(T)$ data of EuNi₂Sb₂ measured at 10 kOe are shown in Fig. 2. In the temperature range of 25–300 K the data could be fitted with the Curie-Weiss law, revealing an effective magnetic moment of $\mu_{\rm eff}=8.03(1)~\mu_{\rm B}/{\rm Eu}$ atom and a Weiss constant of $\theta_{\rm P}=-8.5~{\rm K}$. The effective magnetic moment is in good agreement with the theoretical value for Eu²⁺ of 7.94 $\mu_{\rm B}$ and thus also confirms the composition EuNi₂Sb₂. If one would assume the nickel-deficient phase EuNi_{1.53}Sb₂ one obtains a reduced moment of only 7.79 $\mu_{\rm B}/{\rm Eu}$ atom. The negative $\theta_{\rm P}$ is indicative of antiferromagnetic interactions. At around 5 K antiferromagnetic ordering is evident from the $\chi(T)$ data. To determine the exact Néel temperature $(T_{\rm N})$,

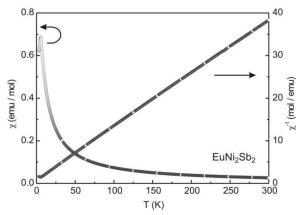


Fig. 2. Temperature dependence of the magnetic susceptibility (χ and χ^{-1} data) of EuNi₂Sb₂ measured at 10 kOe.

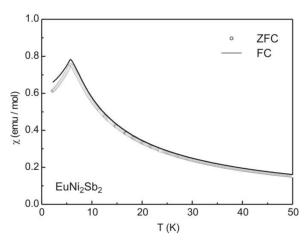


Fig. 3. Susceptibility measurements at 100 Oe of EuNi₂Sb₂ in the zero field-cooled (ZFC) and field-cooled (FC) mode.

low-field susceptibility measurements were performed. The susceptibility in an external field of 100 Oe was first measured in a zero field-cooled and additionally in a field-cooled state in the temperature range of 2.5 – 50 K (Fig. 3). An antiferromagnetic transition at 5.8(1) K was observed. Fig. 4 displays the magnetization isotherms taken at different temperatures. The magnetization isotherm above the ordering temperature at 50 K shows a linear increase of magnetization with the application of an external field, as is expected for a paramagnetic material. Close to the ordering temperature, the isotherm at 10 K reveals a slight curvature at high fields. Below the ordering temperature at 3 K the magnetization isotherm displays a metamagnetic step starting at around 20 kOe. The magnetization isotherm at 3 K shows a saturation magnetization of Note 1181

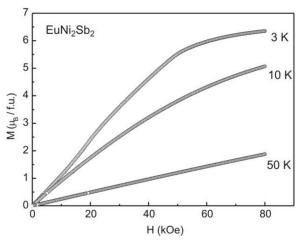


Fig. 4. Magnetization isotherms of $EuNi_2Sb_2$ measured at 3 (below T_N), 10 and 100 K.

approximately 6.4 μ_B /Eu atom at 80 kOe, close to the maximum theoretical value of $g_J \times S = 7 \ \mu_B$ /Eu²⁺. The M(H) behavior below the ordering temperature is non-hysteretic, and together with the observed metamagnetic behavior manifests the antiferromagnetic ground state.

¹²¹Sb and ¹⁵¹Eu Mössbauer spectroscopic characterization

The ¹⁵¹Eu Mössbauer spectra of the EuNi₂Sb₂ sample at 298, 77, and 4.2 K are presented in Fig. 5 together with transmission integral fits. The corresponding fitting parameters are summarized in Table 1. At 298 and 77 K, well above the magnetic ordering temperature, the spectra could be fitted with a single signal at an isomer shift close to 11.3 mm s⁻¹, indicative of purely divalent europium, in agreement with the susceptibility measurements. The experimental line width is slightly increased with respect to the typical value of around 2.3 mm s⁻¹ observed for intermetallic compounds. This may indicate small inhomogeneities in the sample (*vide infra*).

At 4.2 K, in the magnetically ordered regime, magnetic hyperfine field splitting was observed. However, the spectrum could only be well reproduced by a superposition of two spectral components. The main component with 60 % contribution at $\delta = -11.41(4)$ mm s⁻¹ shows a magnetic hyperfine field of 19.1(1) T, typically observed for europium intermetallics [10]. This signal is superimposed by a second contribution of 40 % at $\delta = -11.23(6)$ mm s⁻¹ and a smaller hyperfine field of

Table 1. Fitting parameters of ^{121}Sb and ^{151}Eu Mössbauer spectroscopic measurements of EuNi_2Sb_2 . Numbers in parentheses represent the statistical errors in the last digit. (δ), isomer shift; (Γ), experimental line width; (ΔE_Q), electric quadrupole splitting; (B_h), magnetic hyperfine field. Numbers marked with an asterisk were fixed during the fitting procedure.

Source	T	δ	Γ	$\Delta E_{ m Q}$	B_{h}	ratio
	(K)	$({\rm mm}{\rm s}^{-1})$	(mm s^{-1})	(mm s^{-1})	(T)	(%)
¹²¹ Sb	77	-8.27(7)	3.4(3)	-0.25(3)	-	-
¹⁵¹ Eu	298	-11.32(3)	3.36(8)	0.0*	-	_
	77	-11.19(2)	2.49(9)	-2.70(3)	_	_
	4.2	-11.41(4)	2.6*	3.5(2)	19.1(1)	60
	4.2	-11.23(6)	2.6*	-1.9(4)	12.9(2)	40

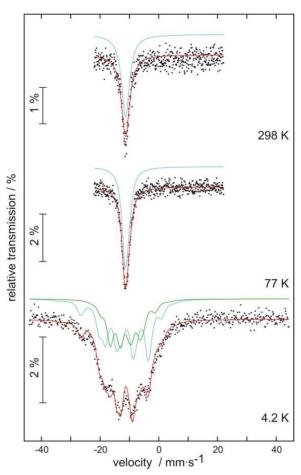


Fig. 5 (color online). Experimental and simulated ¹⁵¹Eu Mössbauer spectra of EuNi₂Sb₂ at 298, 77 and 4.2 K.

only 12.9(2) T. This behavior can only be explained by a domain structure within the sample. The major part of the sample shows full magnetic ordering, while in the other domains long-range magnetic ordering is still

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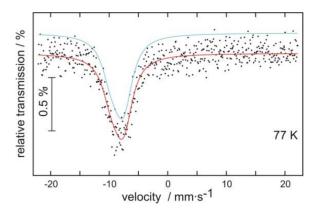


Fig. 6 (color online). Experimental and simulated ¹²¹Sb Mössbauer spectrum of EuNi₂Sb₂ at 77 K.

not complete at 4.2 K. This spectroscopic result indicates small inhomogeneities within the EuNi₂Sb₂ sam-

ple, most likely domains of $EuNi_{2-x}Sb_2$ with a very small value of x. Such domains are most likely also the reason for the slightly enhanced line width observed at r t

The ¹²¹Sb spectrum (Fig. 6) could be well reproduced with a single antimony site. In the fit only a moderate quadrupole splitting of around $-0.25 \,\mathrm{mm \, s^{-1}}$ occurs. The ¹²¹Sb isomer shift compares very well with the ones determined for other intermetallic antimony compounds such as RET_2Sb_2 (RE = Eu, Yb; T = Zn, Mn, Pd) [1,3] or YbTSb (T = Ni, Pd, Pt, Cu, Ag, Au) [11]. Moreover, it is in line with those given in an overview of different antimony compounds by Lippens [12].

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^[1] I. Schellenberg, M. Eul, R. Pöttgen, Z. Naturforsch. 2010, 65b, 18.

^[2] F. Weber, A. Cosceev, A. Nateprov, C. Pfleiderer, A. Faißt, M. Uhlarz, H. von Löhneysen, *Physica B* 2005, 359–361, 226.

^[3] I. Schellenberg, M. Eul, W. Hermes, R. Pöttgen, Z. Anorg. Allg. Chem. **2010**, 636, 85.

^[4] R. Marchand, W. Jeitschko, *J. Solid State Chem.* **1978**, 24, 351

^[5] W. K. Hofmann, W. Jeitschko, J. Less-Common Met. 1988, 138, 313.

^[6] D. Kußmann, R. Pöttgen, U. Ch. Rodewald, C. Rosenhahn, B. D. Mosel, G. Kotzyba, B. Künnen, Z. Naturforsch. 1999, 54b, 1155.

^[7] R. Pöttgen, Th. Gulden, A. Simon, GIT Labor-Fachzeitschrift 1999, 43, 133.

^[8] K. Yvon, W. Jeitschko, E. Parthé, J. Appl. Crystallogr. 1977, 10, 73.

^[9] E. Parthé, L. Gelato, B. Chabot, M. Penzo, K. Cenzual, R. Gladyshevskii, TYPIX – Standardized Data and Crystal Chemical Characterization of Inorganic Structure Types, Gmelin Handbook of Inorganic and Organometallic Chemistry, 8th ed., Springer, Berlin, 1993

^[10] R. Pöttgen, D. Johrendt, Chem. Mater. 2000, 12, 875.

^[11] R. Mishra, R. Pöttgen, R. -D. Hoffmann, Th. Fickenscher, M. Eschen, H. Trill, B. D. Mosel, Z. Naturforsch. 2002, 57b, 1215.

^[12] P.E. Lippens, Solid State Commun. 2000, 113, 399.